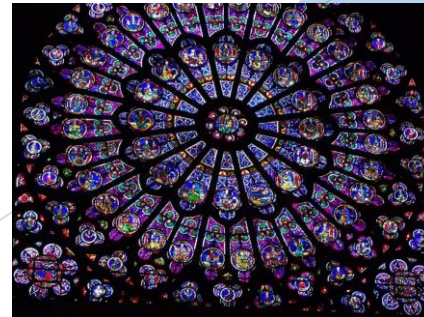
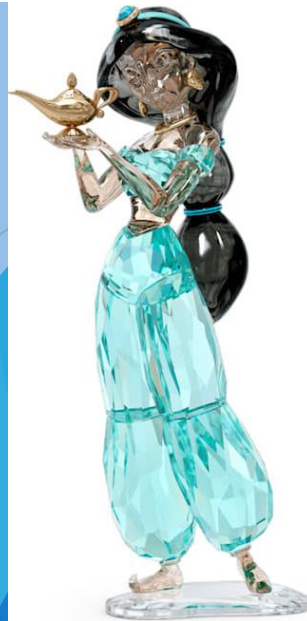
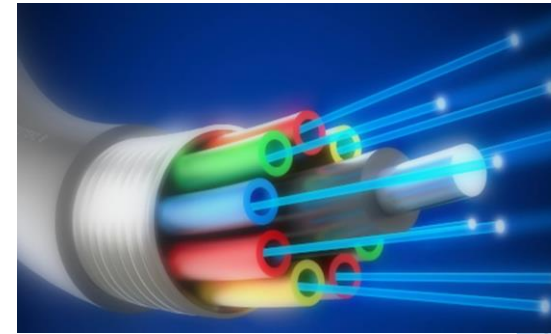
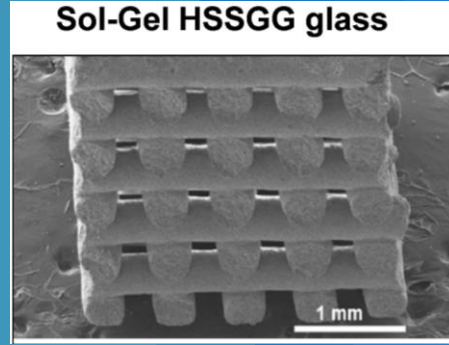
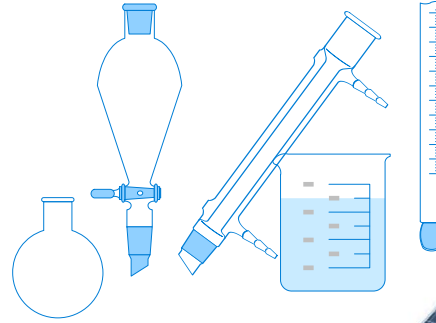


## Sticla, un material cu proprietăți acordabile

Gabriela DANILIUC, Cornelia ILIE,  
George Valentin SĂFTOIU, Ludmila MOTELICA,  
Angela SPOIALA, Denisa FICAI, Anton FICAI



# Sticla: Aplicatii clasice si Emergente



# Impactul **compozitiei** si a modificarii suprafetei



# Tehnici de modificare a suprafetei

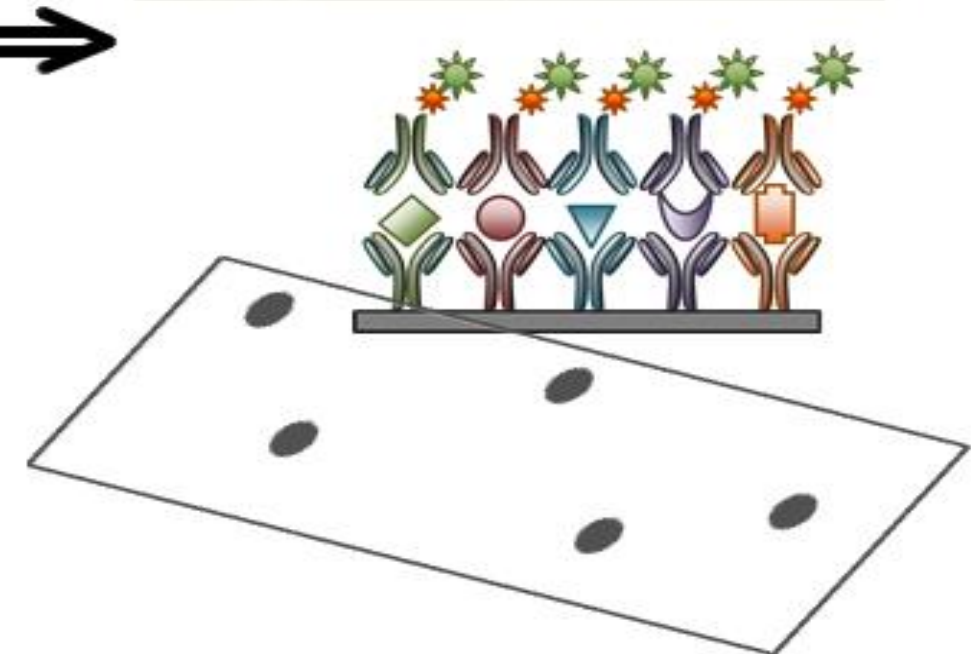
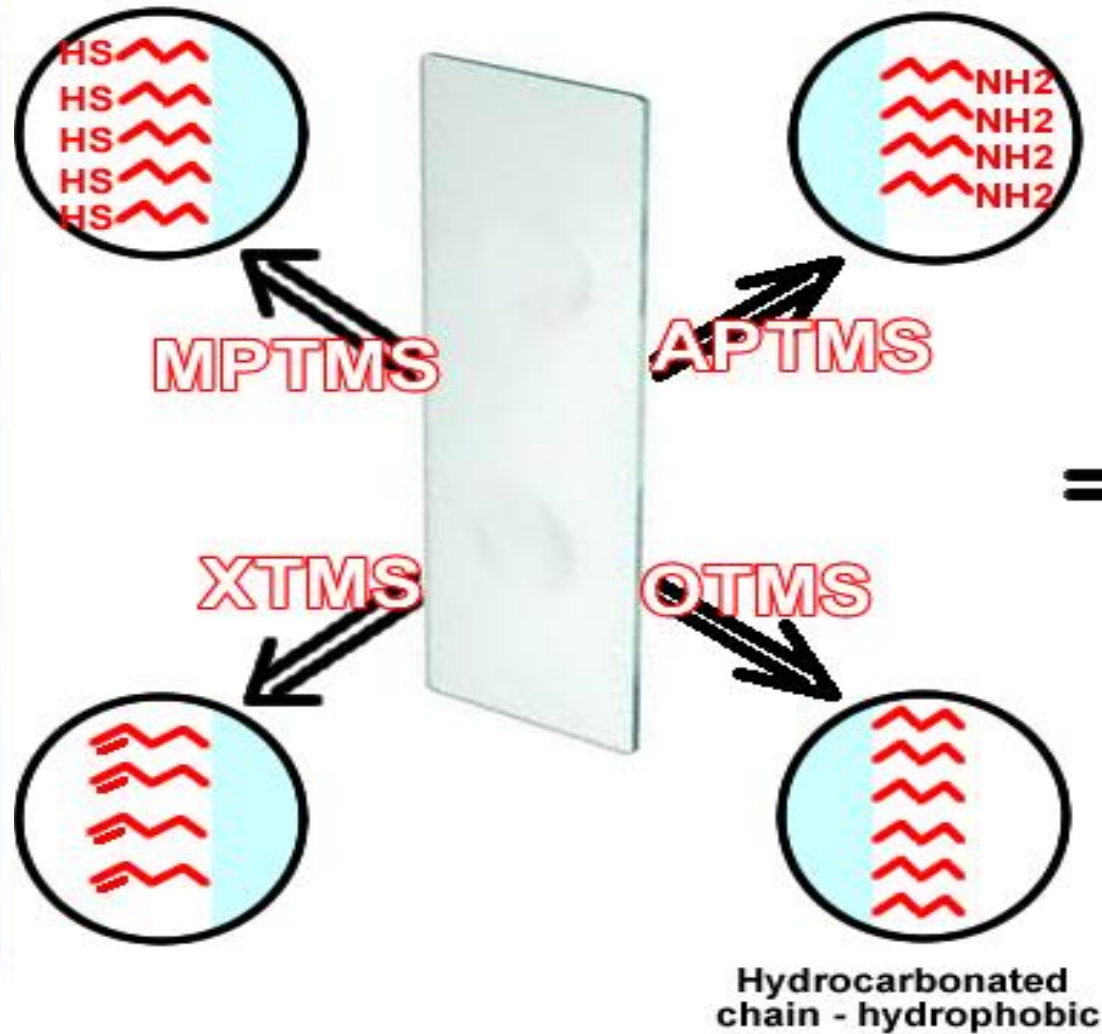
Modificare chimica  
a suprafetei

Modificarea suprafetei

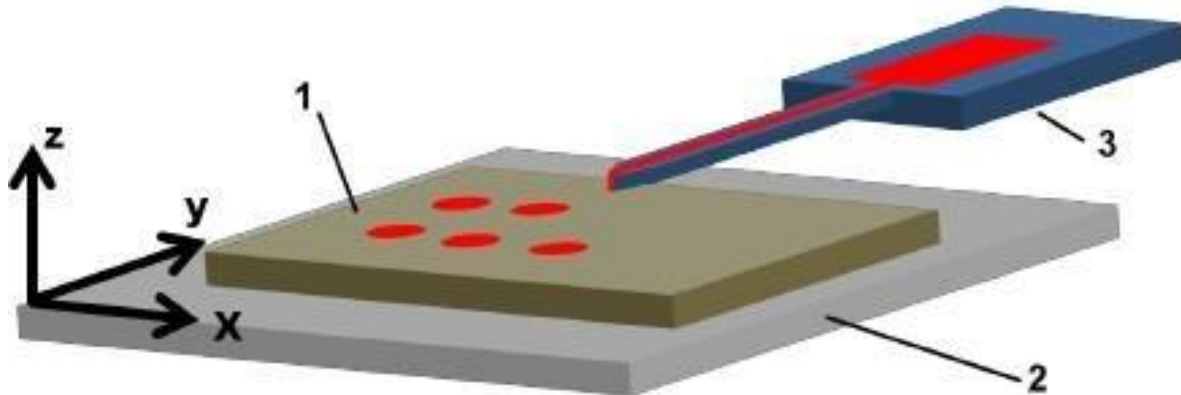
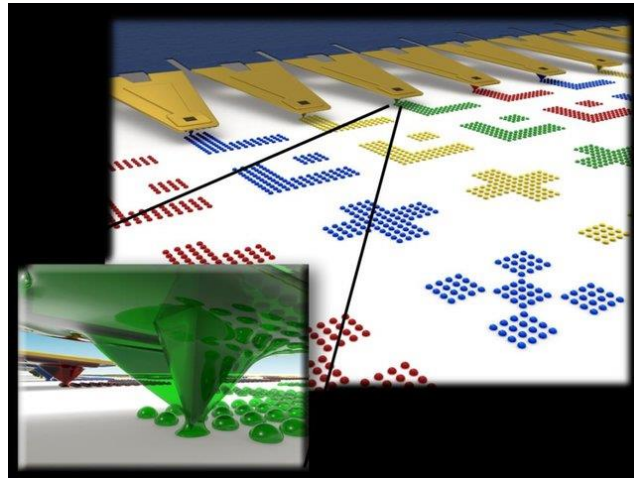
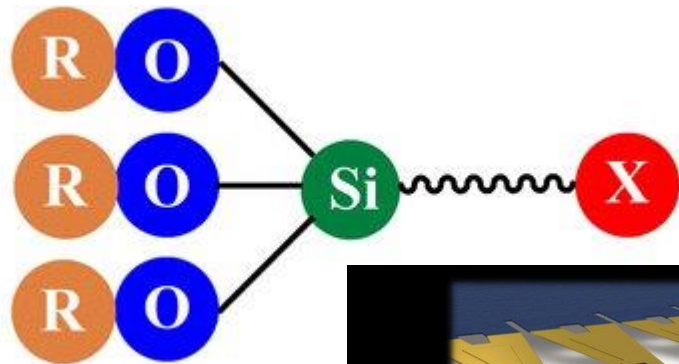
Modificare fizica  
a suprafetei

**Micro and nano-  
structurarea  
(suprafetei)**

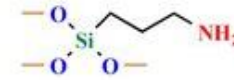
# Functionalizarea chimica a suprafetei vs. aplicatie si performante



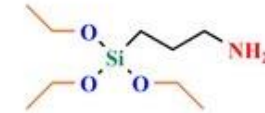
# Agenti de silanizare si tehnici de tratare



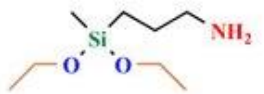
## Aminosilane



(3-Aminopropyl)trimethoxysilane

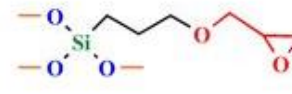


(3-Aminopropyl)triethoxysilane

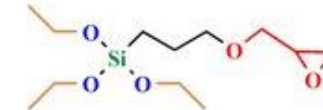


(3-Aminopropyl)diethoxymethylsilane

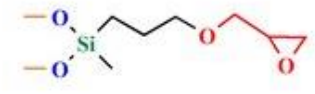
## Glycidoxysilane



(3-Glycidoxypropyl)trimethoxysilane

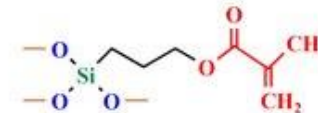


(3-Glycidoxypropyl)triethoxysilane

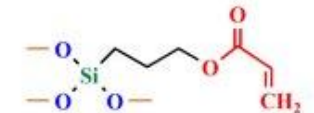


(3-Glycidoxypropyl)dimethoxymethylsilane

## (Meth)acryloxysilane

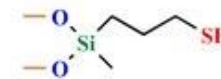


(3-Methacryloxypropyl)trimethoxysilane

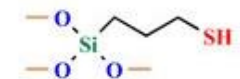


(3-Acryloxypropyl)trimethoxysilane

## Mercaptosilane

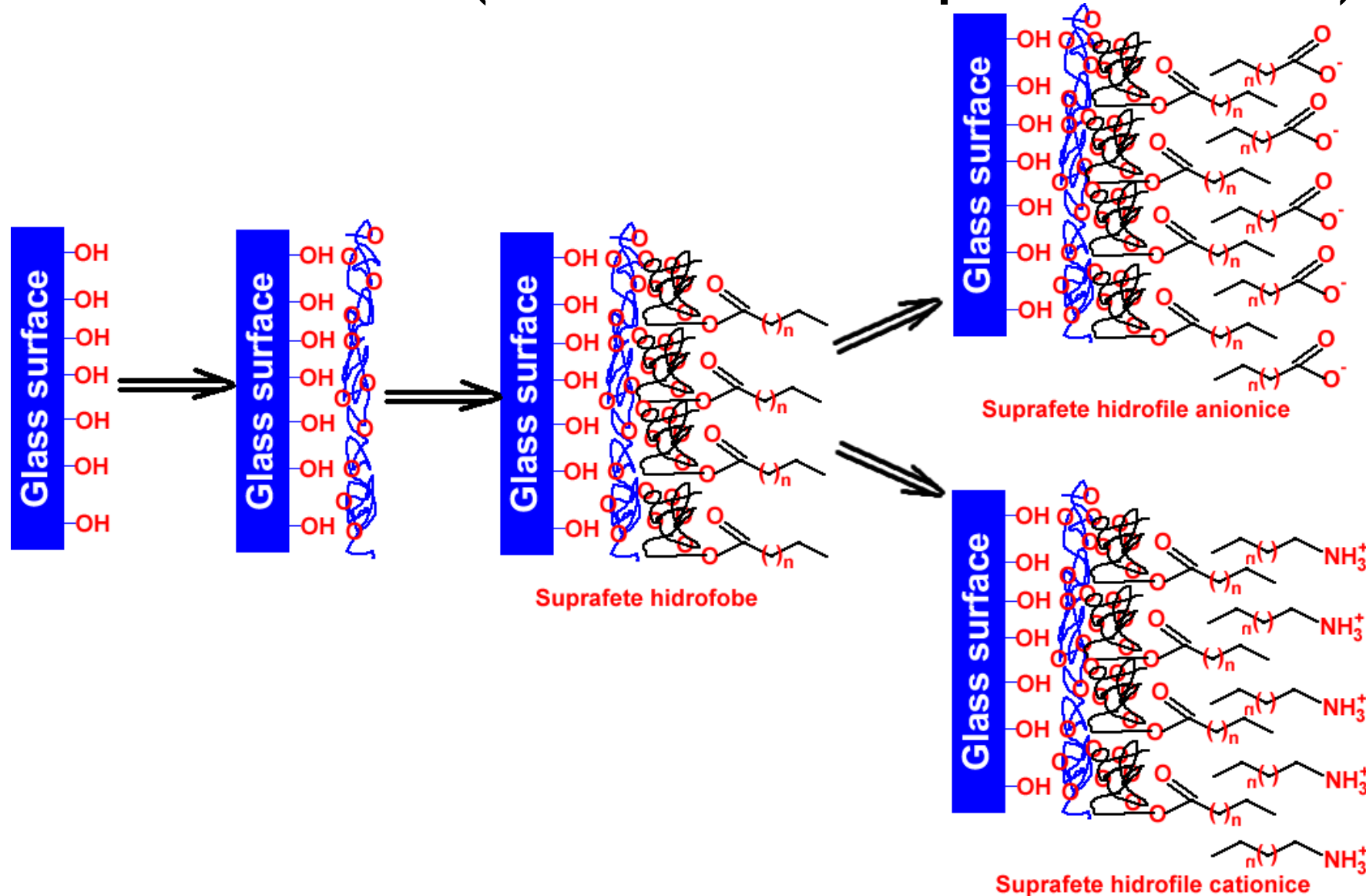


(3-Mercaptopropyl)dimethoxymethylsilane

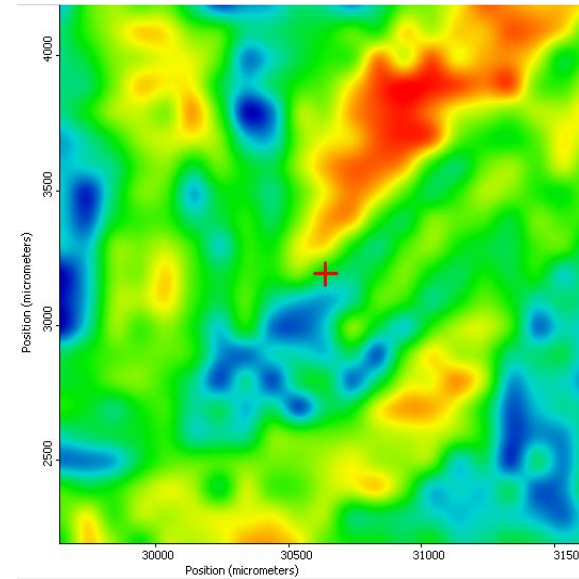
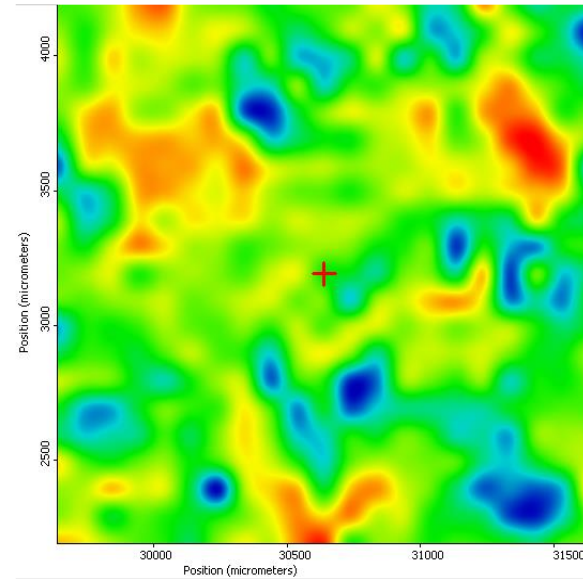


(3-Mercaptopropyl)trimethoxysilane

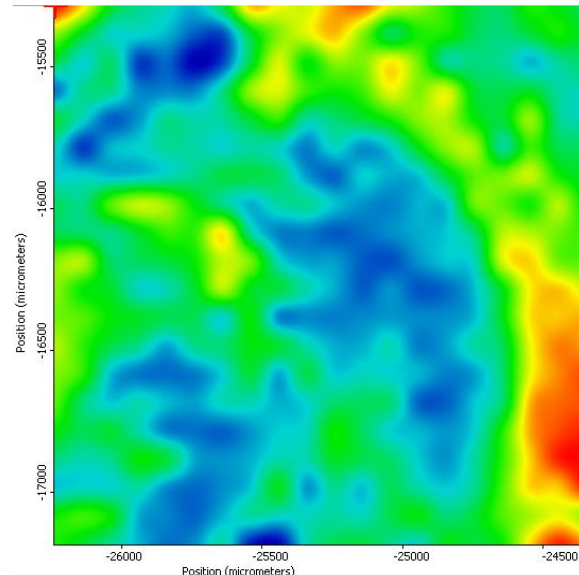
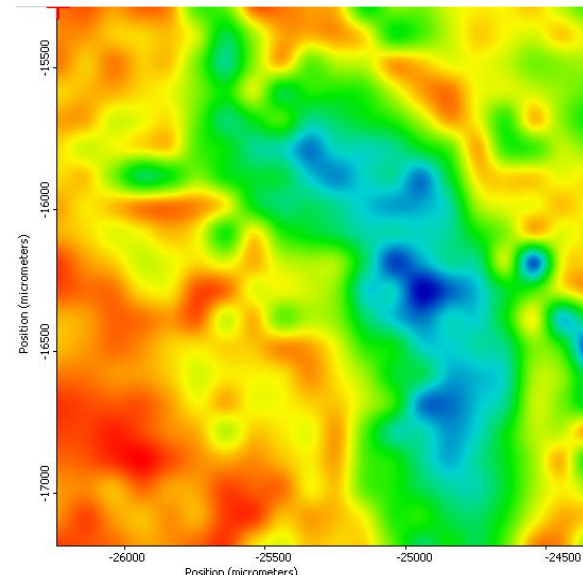
# Modificarea suprafeței de sticlă prin autoasamblare (având la bază depunere de PEG)



# Modificarea complexa a suprafetelor de sticla prin autoasamblare



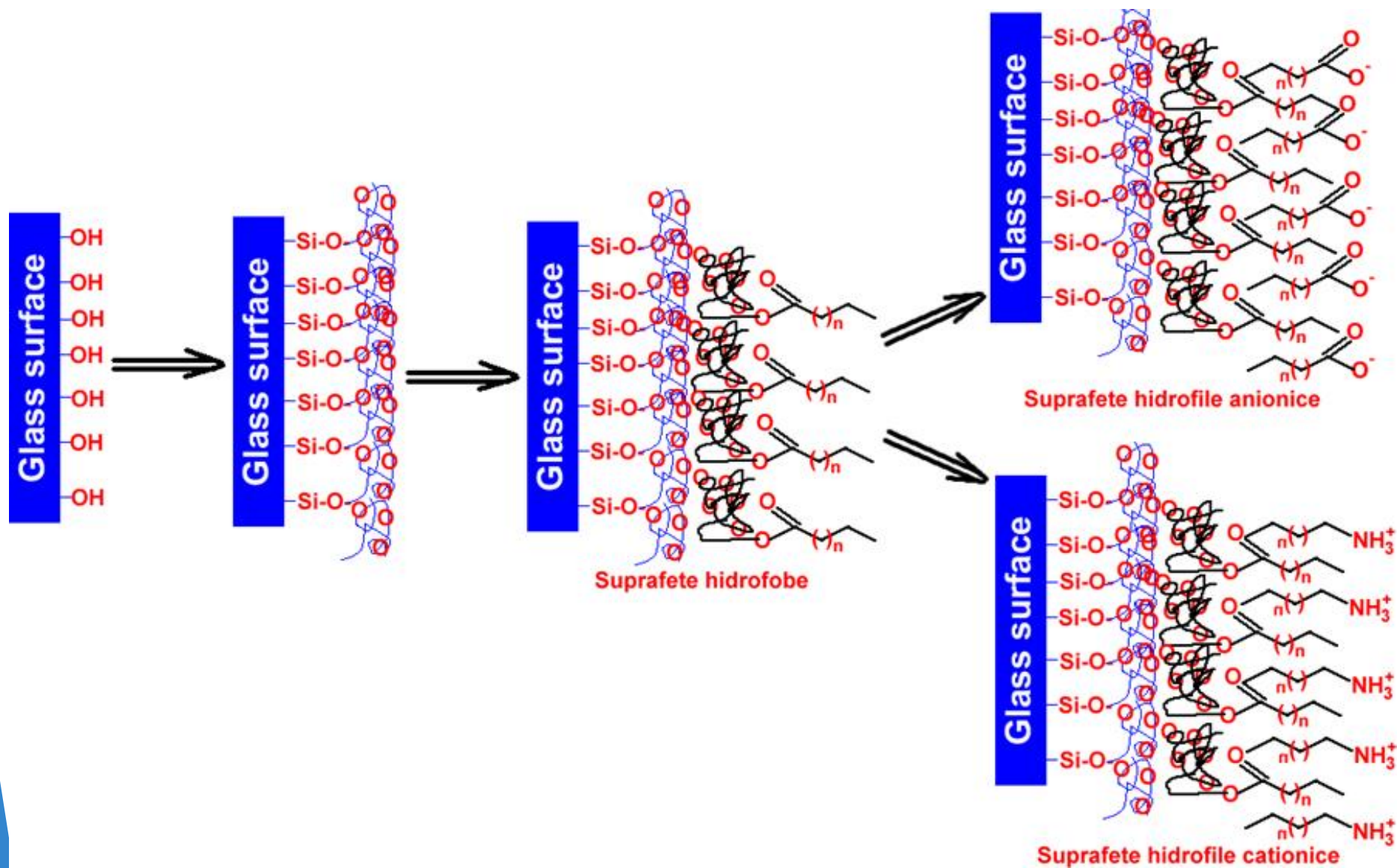
PEG 35 000



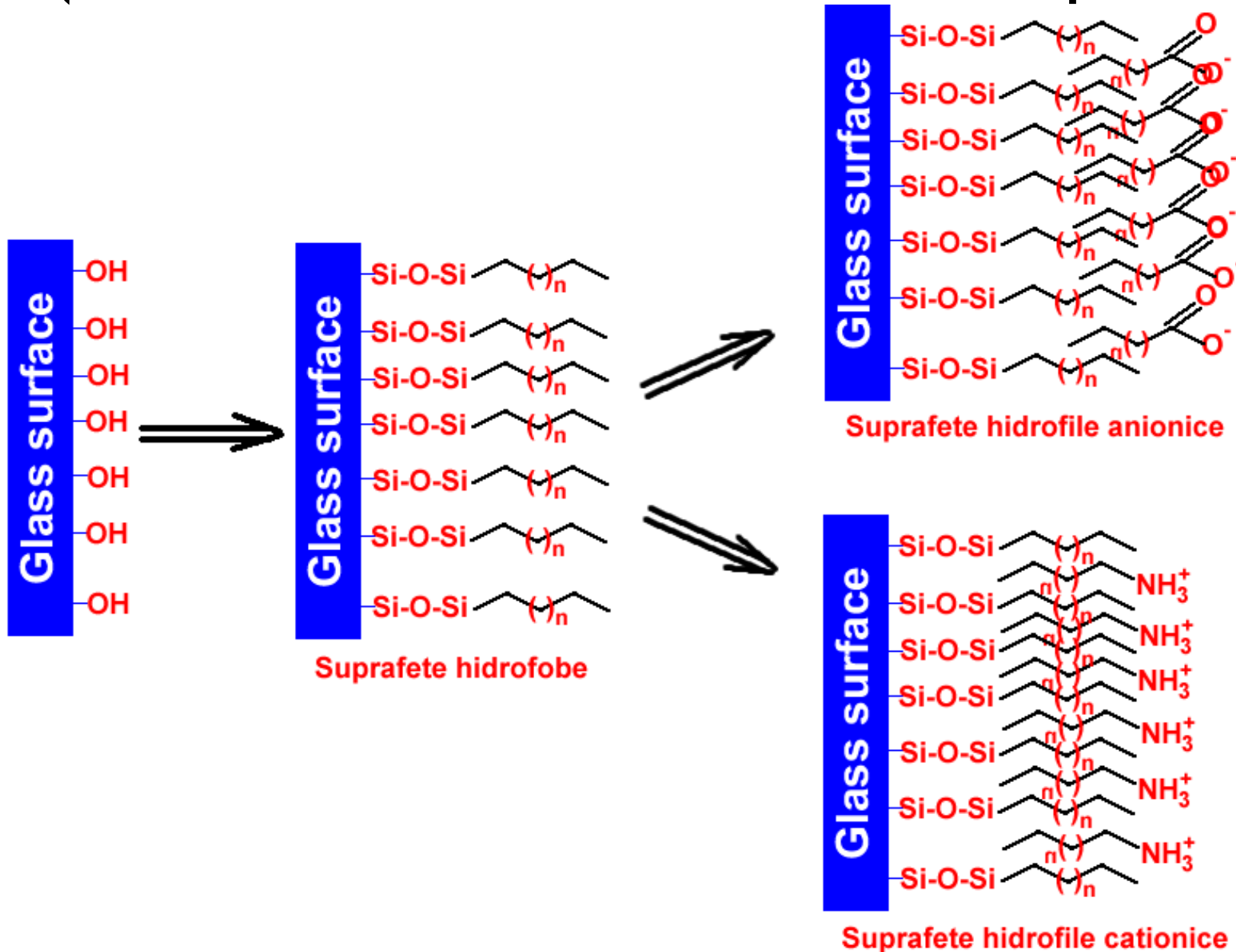
PEG 35000 / Triton X-100

Imagini SEM aferente suprafetelor modificate

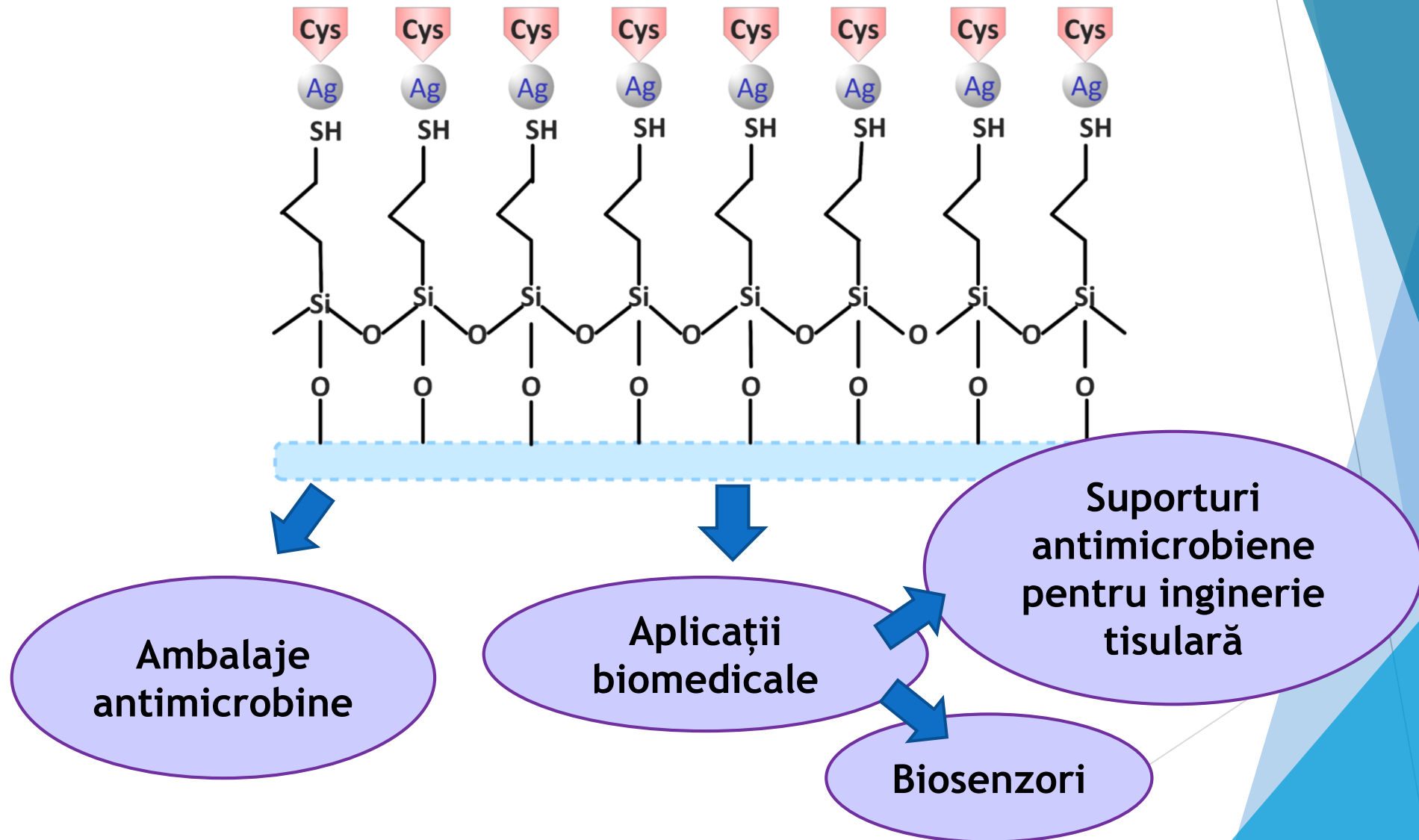
# Modificarea suprafeței de sticlă prin autoasamblare (având la bază silanizare / modificare preliminară chimică)



# Modificarea suprafeței de sticlă prin autoasamblare (având la bază silanizare / modificare preliminară chimică)



# Silanizare suprafeței sticlei cu (3-mercaptopropil)trimetoxisilan și decorarea cu nanoparticule de argint și compuși organici



# Concluzii si perspective

- Sticla, si materialele silicaticе/siloxanice nu sunt material / suprafete inerte chimic;
- Caracteristicile sticlei pot fi induse de compozitie, existand sticle inerte dar si bioactive, cristalinitate controlata, rezistenta termica variabila, etc.;
- Suprafata sticlei poate fi modificata atat fizic cat si chimic, modificarile chimice au avantajul major ca sunt stabile in timp, chiar si in conditii severe;
- Silanizarea este metoda cea mai comuna de modificare a suprafetei materialelor silicaticе, grupele functionale introduce avand rolul de a induce proprietatile noi sau imbunatatite dorite;
- Marea variabilitate a acestor agenti de silanizare permite acordarea proprietatilor cu aplicatiile dorite;
- Modificarea strat cu strat, prin autoasamblare, confera flexibilitate maxima, inducand modificari majore ale suprafetelor: hidrofob=> hidrofil; absorbtie variabila de specii chimice; suprafete aderente/non-aderente; etc.
- Ca perspective ne propunem modificari avansate a suprafetelor in scopul dezvoltarii de suprafete adsorbante (pentru pesticide, antibiotic, material genetic); suprafete aderente/antiaderente (amprente); etc.



## Va multumesc pentru atentie!

CA20126 - Network for research, innovation and product development on porous semiconductors and oxides (NETPORE)

PN-III-P2-2.1-PED-2021-1788, within PNCDI III; 647PED/2022, Nanostructured innovative nutraceuticals with synergistic bioactivities for hepatodigestive protection - Nutrasinpro

TE95/2022: “Funcționalizarea și decorarea cu nanoparticule a suprafeței de sticlă: O abordare promitatoare de a induce aplicații noi” - Glass.

**Intrebari?!**

